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Based on the examples provided, which set(s) of reactions in Model 1 typically involve ions in solution (A, B, C or D)?

1\ - -l-\ \- -Lj Based on the examples provided, which set(s) of reactions in Model 1 typically involve gases and/or solids? R r!-/ 2 3
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monatomic ions, propose an explanation for the net charge on a polyatomic ion. Your answer should include a discussion of subatomic particles. Somewhere in the ion there are either extra (in the case of an anion) or fewer electrons (in the case of a cation) compared to the total number of protons in the atoms involved. 7.

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compounds formed
from the ions listed
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BaF₂, barium fluoride

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Mg. MgS. magnesium
sulfide Li. Cu . Cu₂O.

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solution?] $\text{pH} = 2.60$ Y

$[\text{H}_3\text{O}^+] = 2.5 \times 10^{-3} \text{ M}$

We can assume that the acid is the major source of hydronium ion and that water contributes little to the total $[\text{H}_3\text{O}^+]$. This concentration of hydronium ion ...

compare to the answer you obtained in Key Question 8 ...

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of 4 If we investigate
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why nothing appears to
happen. First, write all
of the compounds in
the equations, showing
all the ions present
when the reaction is
carried out in water.
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6. Use the table of ions in Model 1 to answer the following questions:

a. In the compound zinc phosphide, what is the charge on the zinc ion?

b. In the compound zinc phosphide, what is

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the charge on the phosphide ion? 7. Explain why a 3 to 2 ratio of ions is necessary for the compound zinc phosphide. POGIL™ Activities for High School Chemistry

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