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Atomic Spectra Flinn Chem Topic

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In this demonstration, this color will be viewed through a Flinn C-Spectra™. It is a holographic diffraction grating that separates, or diffracts, light in the same manner as a traditional ruled diffraction grating. It is ideal for the simple observation of spectral emission lines that make up the observed colors of the glowing gases.

Atomic Emission Spectra - Flinn

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When a pure atomic gas such as hydrogen or helium is subjected to a high-voltage electrical discharge, light is produced and the gas glows. When this light is observed through a diffraction grating, a series of bright colored lines is observed. The series of bright lines is called an atomic emission spectrum and is unique to each element.

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To know the relationship between atomic spectra and the electronic structure of atoms. The concept of the photon, however, emerged from experimentation with thermal radiation, electromagnetic radiation emitted as the result of a source's temperature, which produces a continuous spectrum of energies.

5: Atomic Spectra - Chemistry LibreTexts

Atomic Emission Line Spectra. In a gas discharge tube energy is added to a gas which is adsorbed by electrons. These excited electrons enter high energy orbitals and then fall back to their lower energy and can give off a photon of light of a specific wavelength that can be seen by a diffraction grating or prism.

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